

SOP No.:

Title: Iodine 0.1 M

# PHARMA DEVILS

## **QUALITY CONTROL DEPARTMENT**

GENERAL TESTING PROCEDURE					
	Department:	QC			

<b>Effective Date:</b>		<b>Review Date:</b>	
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Supersede SOP No.:	Nil		

#### 1.0 **OBJECTIVE:**

**1.1** To lay down a procedure for the preparation and standardisation of 0.1 M Iodine.

### 2.0 SCOPE:

**2.1** It is applicable for the estimation of Raw material, bulk product, intermediate product and finish products.

#### 3.0 **RESPONSIBILITY:**

- 3.1 Analyst / Officer / Executive follow the procedure.
- **3.2** Head-QC are responsible for effective implementation of this SOP.

### 4.0 REFERENCE:

**4.1** BP

#### 5.0 **DEFINITION:**

**5.1** Molarity is the number of mole of substance that are present in the given Volume of the solution.

#### **6.0 PROCEDURE:**

## **6.1** Material and Equipment:

**6.1.1** Volumetric flask 1000 ml, Iodine, Potassium iodide, Methyl orange conical flask, record book etc.

## 6.2 Preparation:

**6.2.1** Dissolved 25.4 g of Iodine and 40 g of Potassium Iodide in water and dilute to 1000 ml with the same solvent.

#### **6.3** Standardisation:

**1.1.1** To 10 ml of the iodine solution add 1 ml of dilute acetic acid and 50 ml of water. Titrate with 0.1 M sodium thiosulfate, determining the end-point potentiometrically or using starch solution as indicator.



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1.1 Calculation:  $N_1V_1=N_2V_2$ 

## 7.0 Annexures:

Annexure-I: Molarity Calculation format of Volumetric Solution 0.1 M Iodine.

## 8.0 Distribution:

Display copy 1: Instrument Lab

## 9.0 Abbreviation:

GTP : General Test Procedure

QC : Quality Control laboratories

# 10.0 Revision History:

**Revision history table:** 

<b>Document Number</b>	CC Number/Date	<b>Brief Description of Change</b>



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## **ANNEXURE-I**

# **Molarity Calculation format of Volumetric Solution 0.1 M Iodine**

S. No.	Date	Qty. Prep.	Batch no.	Primary Std. ID. No.	Primary Std. Weight	Calculation	RSD NMT 0.2%	Mean Molarity	Date of Standardization.
1.									
2.									
3.									

Prepared By (Sign/Date): Checked By (Sign/Date):